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APPLICATION NO.	FILING DATE	FIRST NAMED INVENTOR	ATTORNEY DOCKET NO.	CONFIRMATION NO.
10/577,340	01/23/2007	Michael Anthony Wood	KPT 1100	4895
321 SENNIGER PO	7590 11/20/2007 DWERS	EXAMINER		
ONE METROF	POLITAN SQUARE	PUTTLITZ, KARL J		
	ST LOUIS, MO 63102		ART UNIT	PAPER NUMBER
			1621	
•			NOTIFICATION DATE	DELIVERY MODE
			11/20/2007	ELECTRONIC

Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

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•		Application No.	Applicant(s)	
		10/577,340	WOOD ET AL.	
	Office Action Summary	Examiner	Art Unit	
		Karl J. Puttlitz	1621	
Period fo	The MAILING DATE of this communication app	ears on the cover sheet with the c	orrespondence address	
A SHO WHIC - Exter after - If NO - Failur Any r	ORTENED STATUTORY PERIOD FOR REPLY THEVER IS LONGER, FROM THE MAILING DATE IS IN A STATUTORY PERIOD FOR REPLY IS IN A STATE IN A STA	ATE OF THIS COMMUNICATION 16(a). In no event, however, may a reply be timular apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).	
Status				
2a) <u></u>	Responsive to communication(s) filed on <u>28 Ap</u> This action is <b>FINAL</b> . 2b) This Since this application is in condition for allowan closed in accordance with the practice under <i>E</i>	action is non-final. nce except for formal matters, pro		
Dispositi	on of Claims			
5)□ 6)⊠ 7)□	Claim(s) <u>1-20</u> is/are pending in the application.  4a) Of the above claim(s) is/are withdraw Claim(s) is/are allowed.  Claim(s) <u>1-20</u> is/are rejected.  Claim(s) is/are objected to.  Claim(s) are subject to restriction and/or			
Applicati	on Papers			
10)□	The specification is objected to by the Examiner The drawing(s) filed on is/are: a) acce Applicant may not request that any objection to the o Replacement drawing sheet(s) including the correcti The oath or declaration is objected to by the Examiner	epted or b) objected to by the Edrawing(s) be held in abeyance. See on is required if the drawing(s) is obj	e 37 CFR 1.85(a). ected to. See 37 CFR 1.121(d).	
Priority u	nder 35 U.S.C. § 119			
12) Acknowledgment is made of a claim for foreign priority under 35 U.S.C. § 119(a)-(d) or (f).  a) ■ All b) ■ Some * c) ■ None of:  1. ■ Certified copies of the priority documents have been received.  2. ■ Certified copies of the priority documents have been received in Application No. ■  3. ■ Copies of the certified copies of the priority documents have been received in this National Stage application from the International Bureau (PCT Rule 17.2(a)).  * See the attached detailed Office action for a list of the certified copies not received.				
2) Notice	e of References Cited (PTO-892) of Draftsperson's Patent Drawing Review (PTO-948)	4) Interview Summary Paper No(s)/Mail Da	te	
	nation Disclosure Statement(s) (PTO/SB/08) No(s)/Mail Date <u>7/10/2007</u> .	5) Notice of Informal Page 6) Other:	atent Application	

## **DETAILED ACTION**

## Claim Rejections - 35 USC § 103

The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:

(a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.

Claims 1-20 are rejected under 35 U.S.C. 103(a) as being unpatentable over U.S. Patent No. 5,077,442 to Hara et al. (Hara).

The rejected claims cover, inter alia, a homogeneous process for the hydrogenation of dicarboxylic acids and/or anhydrides in the presence of a catalyst comprising: (a) ruthenium, rhodium, iron, osmium or palladium; and (b) an organic phosphine; wherein the hydrogenation is carried out in the presence of at least about 1% by weight water and wherein the reaction is carried out at a pressure of from about 500 psig to about 2000 psig and a temperature of from about 200 C to about 300 C such that from about 1 mol to about 10 mol of hydrogen are used to strip 1 mole of product from the reactor.

With regard to the above embodiments, Hara teaches method for producing 1,4-butanediol by hydrogenating succinic anhydride, succinic acid or .γ-butyrolactone in the presence of a catalyst, wherein the hydrogenation reaction is conducted in a liquid phase using as the catalyst a ruthenium type catalyst comprising ruthenium, an organic phosphine and a phosphorus compound, see column 1, lines 52+.

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The organic phosphine is considered to contribute to the control of the electron state of ruthenium as the main catalyst or to the stabilization of the active state of ruthenium. Specific examples of such an organic phosphine include a trialkylphosphine such as tri-n-octylphosphine, tri-n-butylphosphine or dimethyl-n-octylphosphine, a tricycloalkylphosphine such as tricyclohexylphosphine, a triarylphosphine such as triphenylphosphine, an alkylaryl phosphine such as dimethylphenylphosphine and a polyfunctional phosphine such as 1,2-bis(diphenylphosphino)ethane, see column 2, lines 51+.

Hara also teaches that when succinic anhydride or succinic acid is used as the starting material, water will be formed as a byproduct, as the reaction proceeds. If the reaction is conducted while removing such byproduct water out of the reaction system by a reaction mode such as stripping, for example, by continuously circulating hydrogen in the reaction system, a further improved reaction activity can be attained. After completion of the reaction, desired 1,4-butanediol can be obtained from the reaction solution by a usual separating means such as distillation or extraction. The distillation residue can be recycled as a catalyst component to the reaction system, see column 4, line54+.

The reaction temperature is usually from 50-250 C, and when the reaction is conducted on an industrial scale, the hydrogen pressure in the reaction system is usually from 0.1 to 200 kg/cm<sup>2</sup>, see column 4, lines 39+.

The difference between the process disclosed in Hara and the process covered by the rejected claims is that Hara fails to explicitly teach that about 1 mol to about 10

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mol of hydrogen are used to strip 1 mole of product from the reactor. However, given the pressure ranges of hydrogen disclosed by Hara, those of ordinary skill would expect that the claimed molar ratio of hydrogen to product would covered by the disclosed process, and thus, prima facie obvious.

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## Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Karl J. Puttlitz whose telephone number is (571) 272-0645. The examiner can normally be reached on Monday to Friday from 9 a.m. to 5 p.m.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Yvonne Eyler, can be reached at telephone number (571) 272-0871. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free).

/Karl Puttlitz/
Karl Puttlitz
Primary Examiner
Art Unit 1621